RECEIVED 5616596313 CENTRAL FAX CENTER

U.S. Application No.: 10/826,582

RESPONSE D

Reply to Office action dated 11/27/2006

FEB 2 2 2007

ATTORNEY DOCKET NO.: 3926.081

<u>REMARKS</u>

Claims 21-40 are pending in the application. Claims 1-20 have been previously cancelled.

Claims Rejections - 35 USC 103

Claims 21-22, 25-27, and 30-39 are rejected under 35 USC 103(a) as being obvious over Langer et al. (US 6,155,331) in view of Kriechbaum et al. (US 6,165,926).

Claims 23-24 and 40 are rejected under 35 USC 103(a) as being obvious over Langer et al. in view of Kriechbaum et al. and further in view of either Zoia et al. (US 6,609,043) or Smith et al. (US 6,354,362).

Claims 28-29 are rejected under 35 USC 103(a) as being obvious over Langer et al. in view of Kriechbaum et al. and further in view of Kington (US 4,989,667).

Basically, the Examiner has replaced previously cited Goldsmith with Kriechbaum et al.

As already discussed in the response to the previous Office action, Langer et al. disclose a ceramic casting mold, which is produced by a Rapid process and is in a <u>not-sintered</u> green state. The ceramic consists of molding sand, which is coated with heat-hardened resin binder. The mold strength is provided by the hardened binder, not by the sintering. Clearly, Langer et al. do not disclose

- sintering
- porous ceramic
- fine particles
- · expansion coefficient

T-550 P.003/009 F-632

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From-Akerman Senterfitt

ATTORNEY DOCKET NO.: 3926.081

The Examiner appears to have ignored Applicants' remarks in the response to the previous Office action and has stated again incorrectly that Langer et al. teach about porous ceramic produced by selectively sintering. This is incorrect because Langer et al. teach explicitly a totally different solution "This process is basically different from the known sinter process ..." (see column 13, lines 20 ff). Furthermore, Applicants cannot find anything about porosity, fine particles or expansion coefficient in Langer et al. Maybe that is the reason why the Examiner has not mentioned anything about fine particles or expansion coefficient. It is noted that porosity is not possible with the Langer et al.' process and material because the curable binder fills any space between the filler particles and leaves no room for any porosity.

As already discussed in the response to the previous Office action, the object of the present invention is to guarantee a sufficiently good dimensional stability of the casting mold. This object is achieved by the additional application of fine particles, through which the temperature required for the sintering compound of the coarse particles is lowered, especially shrinkage is reduced.

Langer et al. disclose a totally different alternative solution, namely the compound of the particles, over which the heat-hardened resin binder is coated, is not sintered. This results over the prior art the advantage of reduced material shrinkage and warp (see column 13, lines 27-31).

Applicants have argued in the response to the previous Office action that Langer et al. have already solved the problem of the present invention - but in a different way. So there is no motivation for a person skilled in the art to look for another solution. It is noted that the Examiner has not made any comments about this argument.

Without motivation, a person skilled in the art would neither search for, nor find or use Kriechbaum et al.

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Another reason why a person skilled in the art would not combine Kriechbaum et al. with Langer et al. is that the present invention and Langer et al. concern the production of casting molds with rapid prototyping, whereas Kriechbaum et al. concern a refractory composition and the making of refractory bodies. Accordingly, the IPC and the US Cl. for the two references are totally different.

Further, Applicants believe that a not permitted hindsight consideration presents when considering Kriechbaum et al. To justify this hindsight consideration, the Examiner has argued that Langer et al. fail to teach the use of a castable refractory composition. However, this has nothing to do with the problem or aim of the present invention which is to guarantee a sufficiently good dimensional stability of casting molds.

Even if a person skilled in the art would combine Langer et al. and Kriechbaum et al., the present invention would not result because there is no mention anywhere of a suitable expansion coefficient. The Examiner has argued that alumina particles have well known high thermal expansion coefficient of 8-10 x10⁻⁶K⁻¹. This is incorrect because alumina ceramics have expansion coefficient from 3.1 to 3.7 x10⁻⁶K⁻¹ (see the attachment).

Furthermore, the coarse particles with diameters from 1 to 60 mm in Kriechbaum et al. are far too coarse for rapid processes which normally use particles sizes between 50 and 200 μm.

In addition, the Examiner has stated that the fine particles in Kriechbaum et al. have well known lower sintering temperature of 200 °C below the coarse particles. Unfortunately this is not known to Applicants and the Examiner has no provided any citation from which he obtains this knowledge.

Finally, the Examiner is requested to consider all the remarks presented by Applicants to avoid the unnecessary economic burden of Applicants to have to make the same arguments again.

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Feb-22-07

Reply to Office action dated 11/27/2006

ATTORNEY DOCKET NO.: 3926.081

Withdrawal of the rejections and early issuance of a Notice of Allowance are respectfully requested. Should further issues remain prior to allowance, the Examiner is respectfully requested to contact the undersigned at the indicated telephone number.

Date: February 22, 2007

Respectfully submitted,

Youghong Chen

Registration No. 56,150

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Chart of COE's Coelficient of Thermal Expansion

Adoctional Brazing
 Soldering Support

	Material	10 ⁻⁶ in.	10.6 in./in.*/°F 10.5 in./in.*/°C	10 ⁻⁵ in.	/in.*/°C	
		High	Low	H	Low	
What arazing is A'r wood when Do Yeu Think Brazing?	Zinc & its Alloys ^c	19.3	10.8	3.5	1.9	
The Principle* of Joint Design	Lead & its Alloys ^c	16.3	14.4	5.9	5.6	
a & Basii, Steps in Brazing	Magnesium Alloys ^b	16	14	2.8	2.5	
Chart of COE's Coefficient of	Aluminum & its Alloys ^c	13.7	11.7	2.5	2.1	
Thermal Expansion	Tin & its Alloys ^c	13		2,3	•	
# Handy Flux Temperature Chart	Tin & Aluminum Brasses ^c	11.8	10.3	2.1	1.8	
 6 Safety Brazing 7 lbs nouting Controls Chart 	Plain & Leaded Brasses ^c	11.6	10	2.1	1.8	
 Prixing Solutions Class. Particle Comparison Chart 	Silver	10.9		2.0		
	Cr-Ni-Fe Superalloys ^d	10.5	9.2	1.9	1.7	
John J. 128 Styly Boulett	Heat Resistant Alloys (cast) ^d	10.5	6.4	1.9	1.1	
■ Filler Metals	Nodular or Ductile Irons (cast) ^C	10.4	9.9	1.9	1.2	
a Fluxes	Stainless Steels (cast) ^d	10.4	6.4	1.9	1.1	
fg. 886+	Tin Bronzes (cast) ^c	10.3	10	1.8	1.8	
Bauscli と Lomb Company	Austenitic Stainless Steels ^c	10.2	Ф	1.8	1.6	
■ Ford Hator Company	Phosphor Silicon Bronzes ^c	10.2	9.6	1.8	1.7	
■ IM! Cornellus ■ ITT Automotive	Coppers ^c	8 9	ı	1.8	ı	

http://www.lucasmilhaupt.com/htmdocs/brazing_support/everything_about_brazing/materials_comp_chart.html

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1.4	1.6	1.2	•	1.6	1.2	1.4	1.1	1.5	1,4	0.1		1.1	d.t +	3 1.1	æ,		٥, د	- 2	2 1.0				1.0	1.1 -	1.1 1.0
7.7 1.8	9 1.7	6.8 1.7	. 1.7	9 1.7	6.8 1.7	8 1.6	6.3 1.5	8.1 1.5	8 1.5	5.5 1.5	- 1.4	6.3 1.4	5.7 1.4	5.9 1.3	4,3 1.3	- 1.3	4.9 1.3	. 1.2	5.5 1.2	. 1.2	- 1.2	. 1.1	5.8 1.1		5.8
9.8	9.5	9.4	9.3	9.5	9.5	1.6	8.6	8.4	8.3	8.2	7.9	7.9	7.6	7.5	7.5	7.4	7.1		6.5	6.5	6.5	6.4	6.3	6.2	9
Nickel-Base Superalloys ^d	Aluminum Bronzes (cast) ^c	Cobalt-Base Superalloys ^d	Beryllium Copper ⁶	Cupro-Nickels & Nickel Silvers ^c	Nickel & its Alloys ^d	Cr-Ni-Co-Fe Superalloys ^d	Alloy Steels ^d	Carbon Free-Cutting Steels ^d	Alloys Steels (cast) ^d	Age Hardenable Stainless Steels ^d	Colde	Kigh Temperature Steels ^d	Ultra High Strength Steels ^d	Malleable Irons ^c	Titanium Carbide Cermet ^d	Wrought frons ^c	Titanium & its Alloys ^d	Cobalt ^d	Martensitic Stainless Steels ^c	Nitriding Steels ^d	Palkdium ^c	8eryllum ^b	Chromlum Carbide Cermet ^c	Thorlumb	Cardle Chalabox Stoolet
	Fringalkie Saravel MFG. Co.		■ The Trans Company	Wolverine Brass	1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1 1	Lucas Hilliaupt has compiled a list of the	most commonly asked brazing & softering anestrone. If you have a question you	would liked answered, email us at	Estation Mark Action Operations	requently saved Constraints	교육 	On yo have a difficult brazing problem or need just principal the right brazing	product? We're nere to help. E-mai us your harmon mo'rlens or mostims at	nazing prozent or question or info@fucasinitheupt.com.	Lucas-Milhaupt, Inc.	5656 S. Pennsylvania Ave. Cudahy, WJ 53110, USA	Plione (414) 769-6000 Phone (800) 558-3856	Fax (414) 769-1093							

http://www.lucasmilhaupt.com/htmdocs/brazing_support/everything_about_brazing/materials_comp_chart.html

Chart of COE's Coefficient of Thermal Expansion | Lucas Milhaupt, Inc.

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		1.5		4.7	1	•		•		•		3,8		3.3	2.5		3.1	•	•	3.4	•	•	2.7	2.2	
9	5.8	5.5	5,3	5.2	5.1	5.1	6,4	4.8	4,6	4.6	4.3	4,1	4.1	4	3.9	3.8	3.7	3.7	3.6	3.6	3.4	3.1	3.1	2.4	2.2
Gray Irons (cast) ^c	Beryllium Carbide ^d	Low Expansion Nickel Alloys ^c	Beryllia & Thoria ^e	Alumina Cermets ^d	Molybdenum Disilicide ^c	Ruthenium ^b	Platinum ^c	Vanadium ^b	Rhodlum ^b	Tantalum Carbide ^d	Boron Nitride ^d	Columbium & Its Alloys	Titanium Carbide ^d	Steatile ^c	Tungsten Carbide Cermet ^c	Iridium ^b	* Alumina Ceramics	Zircantum Cərbide ^d	Osmium and Tantalum ^b	Zirconium & its Allays ^b	Hafntum ^b	Zirconta	Molybdenum & its Alloys	Silicon Carbide ^e	Tungsten ^b

Chart of COE's Coefficient of Thermal Expansion | Lucas Milhaupt, Inc.

Electrical Ceramics ^c	7	•	4.	•	•
Zircon ^c	1.8	1.3	w.	7	
Boron Carbide ^e	1.7	•	wj.	•	
Carbon and Graphite ^C	1.5	1.3	ų	7.	
(* or mm/mm)	4000	ev lentand	91140		
"Values represent nign ann now suces or a range or cypical by Value at room temperature only.	a saine a	t back			
^c yalue for a temperature range between room temperature and 212-750° F/100-	ı room tem	perature	and 212-75	:0° F/100-	
390° C.					
dyalue for a temperature range between room temperature and 1000-1800°	room tem	perature	and 1000-1	.800°	
F/540-980° C.					
 Value for a temperature range between room temperature and 2200-2875 	room terr	perature	and 2200-2	875	
F/1205-1580°.C.				-	

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